Protein Solutions™

Calcium Storage Protein Calsequestrin

he sarcoplasmic reticulum (SR) plays an essential role in muscle excitation-contraction coupling by regulating the cytosolic free Ca²⁺ concentration. The main SR proteins responsible for this Ca²⁺ regulation is the Ca²⁺ transport ATPase, the Ca²⁺ storage protein calsequestrin (CSQ), and the Ca²⁺ release channel ryanodine receptor.

CSQ, which binds 60-80 Ca²⁺ ions per molecule, represents a ubiquitous, critically important family of Ca²⁺ storage proteins found not only in the SR but also in the endoplasmic reticulum (ER). We have demonstrated a coupling between ion-dependent polymerization of CSQ and its high capacity Ca²⁺ binding.

In order to study the plausible coupling between the Ca^{2+} binding and the polymerization of CSQs, we have for the first time employed Multi-Angle Light Scattering (MALS) for the *exact* measurement of Ca^{2+} dependent polymeric state of CSQs. We have monitored the elution profiles of CSQs in size exclusion chromatography at various Ca^{2+} concentrations using a DAWN EOS (operating at λ =690 nm) light scattering and an Optilab differential refractometer, the combination of which allows the determination of the absolute molecular weight across the chromatographic peaks.

We found that at a fixed K^+ concentration of 300mM, increasing Ca^{2+} concentration results in transitions from monomer to dimer and to tetramer. If the Ca^{2+} concentration is fixed, *increasing* K^+ concentration results in a shift toward monomers, and *decreasing* K^+ concentration generates polymers.

These data are the *first* demonstrating the interplay of monovalent and divalent cations on the polymeric behavior of purified CSQ, strongly suggesting that CSQ polymerization is under dynamic regulation by mono- and divalent cations. Coincidently, Ca²⁺ movement in and out of the SR is always accompanied by the opposing flux of monovalent cations.



Crystal structure of tetrameric calsequestrin

300 mM KCl 0 mM CaCl₂ (b) 300 mM KCl 1 mM CaCl2 (c) 300 mM KCl 3 mM CaClo (d) 70 mM KCl 1 mM CaCl₂ (e) 500 mM KCl Hexamer (H) Tetramer (T) Monomer (M) 12 14 16 20 Volume (ml)

Figure 1. Polymer states of skeletal calsequestrin in different ionic environments; the molecular weights are depicted with different colors.



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